Stereospecific Total Synthesis of (±)-Albene via a Prochiral Precursor

Adusumilli Srikrishna^{*} and Sankuratri Nagaraju Department of Organic Chemistry, Indian Institute of Science, Bangalore 560012, India

The total synthesis of racemic albene **2** *via* the prochiral precursor **3**, using a stereoselective Claisen rearrangement and an intramolecular diazoketone cyclopropanation as key reactions, is described.

The natural product albene was first isolated ¹ in 1962 from *Petasites albus* and the structure, as the *exo* isomer **2**, was established ² conclusively in 1978 after an initial assignment ³ as the *endo* isomer **1** (now referred to as isoalbene). The unique 2,6-dimethyltricyclo[$5.2.1.0^{2.6}$]decane skeleton incorporating two contiguous quaternary carbons makes the compound an interesting synthetic target.⁴ As part of our interest ⁵ in the application of the Claisen rearrangement in the stereospecific synthesis of natural products containing quaternary carbons, we now report a total synthesis of racemic albene **2** *via* the prochiral ketone **3**, using a combination of stereoselective Claisen rearrangement and copper catalysed intramolecular diazoketone cyclopropanation as key reactions.

The synthetic sequence starting from the cyclopentadiene-



methyl tetrolate Diels-Alder adduct 4 is depicted in Scheme 1. Diels-Alder reaction of cyclopentadiene with but-2-ynoic acid followed by esterification with ethereal diazomethane furnished, in 68% yield, the requisite starting material 4.6 The allyl alcohol 5 required for the Claisen rearrangement was obtained, in 90% yield, by selective hydrogenation of the adduct 4 followed by



Scheme 1 i, H₂/10%Pd-C, EtOAc, 2 h; ii, DIBAH, toluene, -78 °C \rightarrow room temp., 1 h; iii, MeC(OEt)₃, EtCO₂H, 180 °C, 48 h; iv, MeOH-H₂O, NaOH, reflux, 6 h; v, (COCl)₂, C₆H₆, room temp., 2 h; vi, CH₂N₂, Et₂O, room temp., 2 h; vii, anhydrous Cu₂SO₄, W-lamp, C₆H₁₂, 5 h; viii, Li, liq. NH₃, THF, 5 min, 80%; or H₂/10%Pd-C, EtOH, 40 psi, 6 h, 100%; ix, TsNHNH₂, EtOH, 80 °C, 4 h; x, BuLi, Et₂O-TMEDA, 0 °C \rightarrow room temp., 6 h.

DIBAH reduction of the resultant dihydro compound. The ortho ester Claisen rearrangement⁷ of the allyl alcohol **5** using triethyl orthoacetate in the presence of a catalytic amount of propionic acid (sealed tube, $180 \,^{\circ}$ C, $48 \,$ h) stereoselectively furnished, in 86% yield, the ester **6** via the sterically preferred *exo* transition state. Hydrolysis of the ester **6** with aq NaOH furnished, in 87% yield, the acid **7**, m.p. $86 \,^{\circ}$ C.* Treatment of the acid chloride, obtained from the acid **7** and oxalyl chloride, with an excess of ethereal diazomethane generated the key diazoketone **8**.* Anhydrous copper sulphate-catalysed decompo-

sition⁸ of the diazoketone 8 in refluxing cyclohexane (tungsten lamp), and intramolecular insertion of the resultant keto carbene into the *exo*-methylene, gave stereospecifically the cyclopropyl ketone 9.* Regiospecific cleavage of the cyclopropane ring in 9, using lithium in liquid ammonia, furnished cleanly the symmetric ketone 3, in 80% yield.* Alternatively, catalytic hydrogenation (10%Pd-C, EtOH) of the cyclopropyl ketone 9 quantitatively yielded the prochiral ketone 3. The prochiral ketone 3 was transformed to albene 2 *via* its tosyl hydrazone. Thus, treatment of the ketone 3 with tosylhydrazine furnished the tosyl hydrazone, m.p. 162 °C (decomp.). The tosyl hydrazone on reaction with an excess of BuLi in Et₂O– TMEDA, furnished, in 65% yield, albene 2, m.p. 110 °C (after sublimation) (lit.,³ 110–115 °C); the ¹H NMR spectral results were identical with those reported in the literature.²

In conclusion, we have described here a stereospecific route to albene, *via* the prochiral ketone **3**. A recent report⁹ on the conversion of similar prochiral ketones into chiral alkenes [eqn. (1)], enhances the significance of the present sequence as a potential chiral route to albene 2.



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^{*} Selected spectral data (J values in Hz). For 7: (Found: C, 73.4; H, 9.08. $C_{11}H_{16}O_2$ requires C, 73.3; H, 8.95); $v_{max}(Nujol)/cm^{-1}$ 3000br, 1707, 1240, 935 and 888; δ_H(90 MHz, CDCl₃): δ 4.83 (1 H, s), 4.57 (1 H, s), 2.79 (1 H, br s), 2.48 (1 H, br s), 2.48 and 2.27 (2 H, AB q, J 16), 1.3–1.8 (6 H, m) and 1.2 (3 H, s); δ_c(22.5 MHz, CDCl₃) 179.0 (s), 164.5 (s), 101.3 (t), 46.9 (d), 45.5 (t), 45.0 (d), 44.2 (s), 37.3 (t), 29.3 (t), 23.7 (t) and 23.1 (q). For 8: $v_{max}(film)/cm^{-1}$ 3075, 2105, 1635 and 880; $\delta_{H}(60 \text{ MHz}, \text{CCl}_4)$ 5.08 (1 H, s), 4.65 (1 H, s), 4.41 (1 H, s), 2.66 (1 H, br s), 2.5 (1 H, br s), 2.2 (2 H, ABq, J 14), 1.3-1.9 (6 H, m) and 1.13 (3 H, s). For 9: v_{max}(film)/cm⁻¹ 1730, 1205 and 957: δ_H(90 MHz, CDCl₃) δ 2.48 and 2.02 (2 H, ABq, J 18), 2.16 (1 H, br s), 1.2-2.0 (10 H, m) and 0.98 (3 H, s); δ_c(22.5 MHz, $CDC1_3$) 215.1 (s), 56.0 (t), 50.8 (s), 48.2 (d), 42.0 (d), 40.5 (s), 37.7 (t), 35.2 (d), 25.0 (2 C, t), 23.1 (q) and 17.2 (t) (Found: M⁺, 176.1195. $C_{12}H_{16}O$ requires M, 176.1201). For 3: $v_{max}(film)/cm^{-1}$ 1735, 1465, 1405, 1385, 1090 and 625; δ_H(90 MHz, CDCl₃) 2.32 (4 H, s), 1.9 (2 H, br s), 1.2–1.8 (6 H, m) and 1.12 (6 H, s); δ_C(22.5 MHz, CDCl₃) 219.5 (s), 55.4 (2 C, t), 50.5 (2 C, d), 46.2 (2 C, s), 34.8 (t), 23.7 (2 C, t) and 21.1 (2 C, q) (Found: M⁺, 178.1364. C₁₂H₁₈O requires M, 178.1358).